

Table 2. Selected parameters of the intramolecular hydrogen bond observed in the 4-carbamoyl and 4-butoxycarbonyl analogs

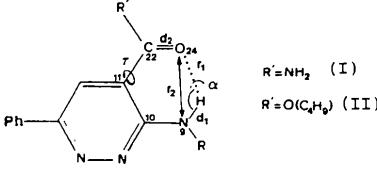
		$R' = \text{NH}_2$ (I)	$R' = \text{O}(\text{C}_2\text{H}_5)_2$ (II)
	'10'	'100'	
d_1 (Å)	1.098 (6)	1.112 (6)	0.980 (3)
d_2 (Å)	1.237 (9)	1.228 (9)	1.200 (4)
r_1 (Å)	1.830 (6)	1.818 (3)	2.004 (3)
r_2 (Å)	2.642 (9)	2.649 (7)	2.722 (4)
α (°)	127.1 (4)	127.7 (4)	128.3 (1)
β (°)	6.6 (12)	7.9 (11)	0.4 (6)
Deviations (Å) from the mean plane of the pyridazine ring			
C(22)	0.008	0.006	0.025
O(24)	0.145	0.162	0.020
N(9)	0.063	0.026	-0.057
H(9)	0.091	0.172	-0.097
E.s.d.'s (Å)	0.013	0.002	0.010

Table 3. Intermolecular hydrogen-bond distances (Å) and angles (°) for compound (I)

N...H-O	N...O	H-O	N...H	$\angle N-H-O$
N(4')...H(992)-O(99 ⁱⁱ)	2.877 (7)	1.176 (4)	1.734 (6)	162.5 (3)
N(14')...H(991)-O(99 ⁱⁱ)	2.905 (7)	0.954 (4)	2.087 (6)	142.8 (3)
N(104')...H(552)-O(55 ^v)	2.903 (8)	1.083 (4)	1.849 (6)	163.0 (2)
 N-H...N				
N(23')-H(232)...N(115)	2.954 (9)	1.088 (6)	2.026 (6)	141.3 (4)
 N-H...O				
N(23')-H(231)...O(55 ^v)	2.868 (7)	1.067 (6)	1.872 (3)	153.7 (4)
N(123')-H(123)...O(99 ^v)	3.047 (6)	1.089 (6)	1.984 (3)	167.7 (2)

Symmetry code: (i) x, y, z ; (ii) $x, -y + \frac{1}{2}, z - \frac{1}{2}$; (iii) $-x - \frac{1}{2}y, z - \frac{1}{2}$; (iv) $-x, y - \frac{1}{2}, -z + \frac{1}{2}$; (v) $x - \frac{1}{2}, -y + 1, -z + \frac{1}{2}$.

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Structure of a Tricyclo[4.3.0.0^{3,7}]nonane Derivative

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Abstract. 4-(5,8-Diacetyl-6,7-dihydroxy-3-methyltricyclo[4.3.0.0^{3,7}]non-1-yl)-2-butanone, $C_{18}H_{26}O_5$, $M_r = 322.40$, monoclinic, $P2_1/c$, $a = 13.762 (2)$, $b = 10.507 (2)$, $c = 12.110 (1)$ Å, $\beta = 97.338 (8)$ °, $V =$

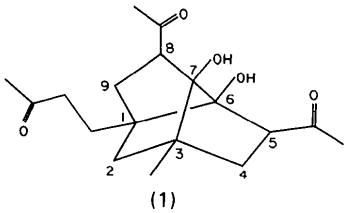
$1736.7 (4)$ Å³, $Z = 4$, $D_x = 1.233$ Mg m⁻³, $Cu K\alpha$, $\lambda = 1.5418$ Å, $\mu = 0.69$ mm⁻¹, $F(000) = 696$, $T = 293$ K, $R = 0.050$ for 2761 unique reflections. It has been confirmed that the compound formed by Michael

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reaction of 3-methyl-1,2-cyclopentanedione with excess methyl vinyl ketone has the novel tricyclononane skeleton. Endocyclic bond angles at positions 1, 3, 6 and 7 are remarkably decreased, 94.0 (1)–100.7 (2) $^\circ$, while the exocyclic angles are increased, 118.2 (1)–116.9 (1) $^\circ$, indicating a considerable strain in the skeleton. The C–C bond lengths lie in a similar range as found in some 8,9,10-trinorbornane rings: 1.529 (1)–1.557 (3) Å. The two six-membered rings take boat conformations, and the three five-membered rings pseudo C_2 conformations.

Introduction. 4-(5,8-Diacetyl-6,7-dihydroxy-3-methyltricyclo[4.3.0.0^{3,7}]non-1-yl)-2-butanone (1) was obtained as a Michael-reaction adduct of 1 mol of 3-methyl-1,2-cyclopentanedione with 3 mol of methyl vinyl ketone under basic conditions (Utaka, Fujii & Takeda, unpublished). Since such a Michael reaction is unprecedented and the structure of the adduct could not be unambiguously determined by spectral analyses (IR, 1H and ^{13}C NMR), X-ray structure analysis of (1) has been carried out, and its novel tricyclo[4.3.0.0^{3,7}]nonane skeleton has been confirmed.



Experimental. Colorless prismatic crystal (slow evaporation from ether solution) elongated along c with developed faces {100}, dimensions 0.30 × 0.28 × 0.35 mm. Rigaku AFC-5 four-circle diffractometer. Lattice parameters determined with 20 reflections in the range $10 < 2\theta < 47^\circ$ by least-squares method; intensities measured up to $\sin\theta/\lambda$ 0.5753 Å⁻¹, ω –2 θ scan method [scan speed 4° min⁻¹ in ω , scan range (2 θ): 1.2° + 0.15° tan θ], Ni-filtered Cu $K\alpha$, 40 kV, 200 mA (rotating anode), background measured for 4 s on either side of the peak. Three standard reflections measured for every 57 reflections, fluctuation within 1.5% in F . Lorentz and polarization corrections; no absorption correction. 3068 reflections measured, all 2761 unique reflections (ranging over $h = -15$ to 15, $k = 0$ to 12, $l = 0$ to 13) used for refinement; 2627 reflections larger than $1.0\sigma(F_o)$. $R_{\text{int}} = 0.010$ for 149 $hk0$ reflections. Structure solved by MULTAN78, and refined (anisotropically for non-H atoms) by block-diagonal least squares; $\sum w(|F_o| - |F_c|)^2$ minimized, with $w = 1.0 / |\sigma(F_o)|^2 - 0.0364|F_o| + 0.0028|F_o|^2$ for $|F_o| > 0$, $w = 1.6983$ for $|F_o| = 0$; H-atom positions determined from difference Fourier map, and refined isotropically by least squares. Extinction correction for ten strongest

Table 1. Final atomic parameters ($\times 10^4$) with e.s.d.'s in parentheses

	x	y	z	B_{eq} (Å ²)
O(1)	3934.6 (8)	5674 (1)	4785.6 (9)	4.43 (5)
O(2)	2112.6 (7)	7387.1 (9)	4635.0 (8)	3.74 (4)
O(3)	1977 (1)	5128 (1)	3255 (1)	5.23 (6)
O(4)	-640 (1)	7871 (1)	7515 (2)	8.58 (9)
O(5)	4360 (1)	8315 (1)	5011 (1)	6.08 (6)
C(1)	3208 (1)	5618 (1)	5510 (1)	3.32 (6)
C(2)	2472 (1)	4538 (1)	5141 (1)	3.64 (6)
C(3)	1517 (1)	5018 (1)	5555 (1)	3.65 (6)
C(4)	1868 (1)	6122 (1)	6331 (1)	3.15 (6)
C(5)	2611 (1)	5552 (1)	7273 (1)	3.62 (6)
C(6)	3589 (1)	5551 (2)	6769 (1)	3.79 (7)
C(7)	4022 (1)	6883 (2)	6975 (1)	4.36 (7)
C(8)	3262 (1)	7767 (1)	6307 (1)	3.74 (6)
C(9)	2575 (1)	6812 (1)	5622 (1)	3.10 (5)
C(10)	2408 (1)	4365 (1)	3895 (1)	4.19 (7)
C(11)	2910 (2)	3234 (2)	3489 (2)	6.6 (1)
C(12)	1049 (1)	6904 (1)	6719 (1)	3.66 (6)
C(13)	505 (1)	6189 (1)	7535 (2)	4.31 (7)
C(14)	-306 (1)	6901 (2)	7962 (2)	4.98 (8)
C(15)	-715 (1)	6356 (2)	8940 (2)	6.5 (1)
C(16)	4290 (1)	4477 (2)	7147 (2)	5.48 (9)
C(17)	3722 (1)	8670 (2)	5542 (1)	4.49 (8)
C(18)	3357 (2)	9999 (2)	5448 (2)	6.6 (1)

reflections [$I_{\text{corr}} = I_o(1 + 1.29 \times 10^{-5}I_c)$]. $R = 0.050$ for 2761 reflections, $wR = 0.062$, $S = 2.18$, $(\Delta/\sigma)_{\text{max}}$ in final refinement cycle 0.3; max. and min. $\Delta\rho$ in final difference Fourier map 0.30 and -0.25 e Å⁻³; atomic scattering factors from *International Tables for X-ray Crystallography* (1974), computations carried out at the Crystallographic Research Center, Institute for Protein Research, Osaka University, and at the Okayama University Computer Center; programs MULTAN78 (Main, Hull, Lessinger, Germain, Declercq & Woolfson, 1978), HBLS-V (Ashida, 1973), MOLCON (Fujii, 1979) and ORTEP (Johnson, 1965).

Discussion. Final atomic parameters are listed in Table 1.* A projection of the molecular structure is shown in Fig. 1. Bond lengths and angles are listed in Table 2, and torsion angles relevant to the conformation of the tricyclononane ring in Table 3.

The tricyclononane skeleton is constituted of two six-membered rings and three five-membered rings [rings (1) and (2), and rings (3), (4) and (5), respectively, in Table 3]. The endocyclic torsion angles show that both six-membered rings take boat conformations as found in an 8,9,10-trinorbornane group (Léger, Colleter & Carpy, 1982). Overtwisting from the ideal boat form (Hendrickson, 1961) amounts to 24.6° at χ_3 for ring (1) and 24.4° at χ_9 for ring (2). The

* Lists of structure factors, anisotropic thermal parameters, H-atom parameters and bond lengths and angles involving H atoms have been deposited with the British Library Lending Division as Supplementary Publication No. SUP 42566 (17 pp.). Copies may be obtained through The Executive Secretary, International Union of Crystallography, 5 Abbey Square, Chester CH1 2HU, England.

five-membered rings take pseudo C_2 conformations, differing from the C_s conformation found in the 8,9,10-trinorbornane group (Léger *et al.*, 1982). The largest overtwisting from the ideal C_2 conformation (Hendrickson, 1961) is 17.4° at χ_{18} in ring (4).

The C—C bond lengths 1.529 (1)–1.557 (3) Å in the tricyclononane skeleton lie in a similar range to that found in some 8,9,10-trinorbornane rings (Léger *et al.*, 1982; Beckwith, Rodgers & Wagner, 1982). Endocyclic bond angles at C(1), C(4), C(6) and C(9) are remarkably decreased, 94.0 (1)–100.7 (2) $^\circ$, while the exocyclic angles at these atoms are much increased,

118.2 (1)–116.9 (1) $^\circ$. Such a correlation between the decreased internal angles and the increased external angles is also observed in the 5,7,8,9-tetraoxatricyclo[4.2.1.0^{2,6}]nonane ring (Henke & Keul, 1975), indicating a considerable strain in these skeletons. The 5- and 8-acetyl groups take *exo* conformations (χ_{27} and χ_{33} in Table 3, respectively), intramolecular short contacts with the H atoms at 5 and 8 positions being H(2)…C(16) 2.90 (2), H(2)…H(16B) 2.34 (3); H(8)…C(12) 2.88 (1), H(8)…H(12A) 2.32 (2) Å.

The hydroxyl groups at 6 and 7 participate in intramolecular hydrogen bonds: O(1)…O(5) 2.842 (1), O(1)–H(O1) 0.90 (3), H(O1)…O(5) 1.99 (3) Å,

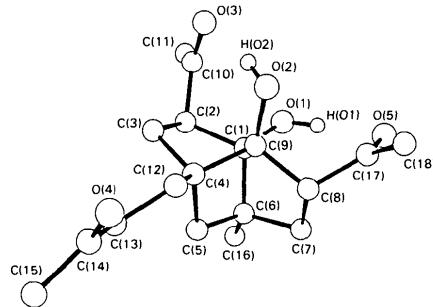


Fig. 1. View of the molecule with numbering of atoms used in the tables. H atoms attached to C atoms are omitted.

Table 2. Bond lengths (Å) and angles ($^\circ$) with e.s.d.'s in parentheses

O(1)–C(1)	1.413 (2)	C(4)–C(12)	1.517 (2)
O(2)–C(9)	1.416 (1)	C(5)–C(6)	1.547 (3)
O(3)–C(10)	1.216 (2)	C(6)–C(7)	1.529 (3)
O(4)–C(14)	1.216 (3)	C(6)–C(16)	1.517 (3)
O(5)–C(17)	1.212 (2)	C(7)–C(8)	1.548 (3)
C(1)–C(2)	1.548 (3)	C(8)–C(9)	1.545 (2)
C(1)–C(6)	1.549 (3)	C(8)–C(17)	1.520 (3)
C(1)–C(9)	1.543 (2)	C(10)–C(11)	1.489 (3)
C(2)–C(3)	1.550 (3)	C(12)–C(13)	1.513 (3)
C(2)–C(10)	1.511 (3)	C(13)–C(14)	1.490 (3)
C(3)–C(4)	1.532 (2)	C(14)–C(15)	1.489 (3)
C(4)–C(5)	1.552 (2)	C(17)–C(18)	1.483 (3)
C(4)–C(9)	1.557 (1)		
O(1)–C(1)–C(2)	109.9 (2)	C(7)–C(6)–C(16)	114.3 (2)
O(1)–C(1)–C(6)	115.8 (2)	C(6)–C(7)–C(8)	103.9 (2)
O(1)–C(1)–C(9)	118.2 (1)	C(7)–C(8)–C(9)	102.5 (1)
C(2)–C(1)–C(6)	112.2 (2)	C(7)–C(8)–C(17)	112.8 (2)
C(2)–C(1)–C(9)	105.4 (1)	C(9)–C(8)–C(17)	110.6 (2)
C(6)–C(1)–C(9)	94.3 (1)	O(2)–C(9)–C(1)	118.1 (1)
C(1)–C(2)–C(3)	102.6 (2)	O(2)–C(9)–C(4)	114.8 (1)
C(1)–C(2)–C(10)	109.2 (2)	O(2)–C(9)–C(8)	110.7 (1)
C(3)–C(2)–C(10)	114.7 (2)	C(1)–C(9)–C(4)	94.0 (1)
C(2)–C(3)–C(4)	103.4 (1)	C(1)–C(9)–C(8)	105.2 (1)
C(3)–C(4)–C(5)	106.6 (1)	C(4)–C(9)–C(8)	112.9 (1)
C(3)–C(4)–C(9)	100.6 (1)	O(3)–C(10)–C(2)	121.3 (2)
C(3)–C(4)–C(12)	114.3 (1)	O(3)–C(10)–C(11)	121.7 (2)
C(5)–C(4)–C(9)	100.7 (1)	C(2)–C(10)–C(11)	117.1 (2)
C(5)–C(4)–C(12)	115.3 (1)	C(4)–C(12)–C(13)	112.5 (2)
C(9)–C(4)–C(12)	117.4 (1)	C(12)–C(13)–C(14)	115.6 (2)
C(4)–C(5)–C(6)	103.6 (1)	O(4)–C(14)–C(13)	121.6 (2)
C(1)–C(6)–C(5)	100.7 (2)	O(4)–C(14)–C(15)	121.1 (2)
C(1)–C(6)–C(7)	101.2 (2)	C(13)–C(14)–C(15)	117.3 (2)
C(1)–C(6)–C(16)	116.9 (2)	O(5)–C(17)–C(8)	121.2 (2)
C(5)–C(6)–C(7)	106.0 (2)	O(5)–C(17)–C(18)	120.7 (2)
C(5)–C(6)–C(16)	115.7 (2)	C(8)–C(17)–C(18)	118.1 (2)

Table 3. Torsion angles ($^\circ$) concerning the conformation of the tricyclononane skeleton

Endocyclic		
(I) Ring (1) through C(1), C(2), C(3), C(4), C(5) and C(6)		
χ_1 C(1)–C(2)–C(3)–C(4)	-12.8 (2)	
χ_2 C(2)–C(3)–C(4)–C(5)	-60.4 (2)	
χ_3 C(3)–C(4)–C(5)–C(6)	84.6 (2)	
χ_4 C(4)–C(5)–C(6)–C(1)	-20.7 (2)	
χ_5 C(5)–C(6)–C(1)–C(2)	-54.7 (2)	
χ_6 C(6)–C(1)–C(2)–C(3)	77.8 (2)	
(II) Ring (2) through C(4), C(5), C(6), C(7), C(8) and C(9)		
χ_7 C(6)–C(7)–C(8)–C(9)	-10.0 (2)	
χ_8 C(5)–C(6)–C(7)–C(8)	-62.9 (2)	
χ_9 C(4)–C(5)–C(6)–C(7)	84.4 (2)	
χ_{10} C(9)–C(4)–C(5)–C(6)	-20.0 (2)	
χ_{11} C(8)–C(9)–C(4)–C(5)	-55.2 (1)	
χ_{12} C(7)–C(8)–C(9)–C(4)	75.6 (2)	
(III) Ring (3) through C(1), C(2), C(3), C(4) and C(9)		
χ_1 C(1)–C(2)–C(3)–C(9)	-12.8 (2)	
χ_{13} C(2)–C(3)–C(4)–C(9)	44.2 (1)	
χ_{14} C(3)–C(4)–C(9)–C(1)	-56.2 (1)	
χ_{15} C(4)–C(9)–C(1)–C(2)	48.9 (1)	
χ_{16} C(9)–C(1)–C(2)–C(3)	-23.5 (2)	
(IV) Ring (4) through C(9), C(5), C(6), C(7) and C(1)		
χ_4 C(5)–C(6)–C(1)–C(9)	-20.7 (2)	
χ_{17} C(5)–C(6)–C(1)–C(9)	53.8 (2)	
χ_{18} C(6)–C(1)–C(9)–C(4)	-65.5 (1)	
χ_{19} C(1)–C(9)–C(4)–C(5)	53.1 (1)	
χ_{10}	-20.0 (2)	
(V) Ring (5) through C(1), C(6), C(7), C(8) and C(9)		
χ_{20} C(7)–C(8)–C(9)–C(1)	-25.5 (2)	
χ_{21} C(8)–C(9)–C(1)–C(6)	49.5 (2)	
χ_{22} C(9)–C(1)–C(6)–C(7)	-55.1 (2)	
χ_{23} C(1)–C(6)–C(7)–C(8)	41.8 (2)	
χ_7	-10.0 (2)	
Exocyclic		
χ_{24} O(1)–C(1)–C(2)–C(3)	-151.9 (2)	
χ_{25} O(1)–C(1)–C(6)–C(5)	178.1 (2)	
χ_{26} C(10)–C(2)–C(3)–C(4)	-131.1 (2)	
χ_{27} C(10)–C(2)–C(1)–C(6)	-160.1 (2)	
χ_{28} C(2)–C(3)–C(4)–C(12)	171.0 (1)	
χ_{29} C(6)–C(5)–C(4)–C(12)	-147.4 (1)	
χ_{30} C(4)–C(5)–C(6)–C(16)	-147.8 (2)	
χ_{31} C(8)–C(7)–C(6)–C(16)	168.4 (2)	
χ_{32} C(6)–C(7)–C(8)–C(17)	-129.0 (2)	
χ_{33} C(4)–C(9)–C(8)–C(17)	-163.9 (1)	
χ_{34} C(7)–C(8)–C(9)–O(2)	-154.1 (1)	
χ_{35} C(5)–C(4)–C(9)–O(2)	176.6 (1)	
χ_{36} O(1)–C(1)–C(2)–C(10)	-29.8 (2)	
χ_{37} C(17)–C(8)–C(9)–O(2)	-33.7 (2)	
χ_{38} O(1)–C(1)–C(9)–O(2)	51.2 (2)	

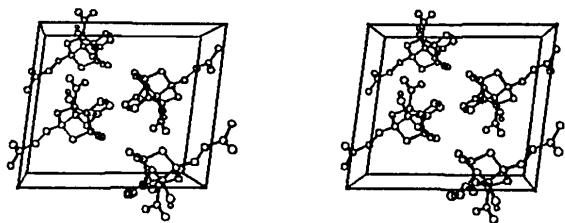


Fig. 2. Stereoscopic view of the molecular packing in the cell. The a axis points from left to right, the b axis onto the plane of the paper, and the c axis downwards. H atoms attached to C atoms are omitted.

$O(1)-H(O1)\cdots O(5)$ 157 (3), $H(O1)\cdots O(5)-C(17)$ 108.3 (9) $^\circ$; $O(2)\cdots O(3)$ 2.888 (2), $O(2)-H(O2)$ 0.82 (2), $H(O2)\cdots O(3)$ 2.10 (2) Å, $O(2)-H(O2)\cdots O(3)$ 164 (2), $H(O2)\cdots O(3)-C(10)$ 104.7 (6) $^\circ$. This finding explains shifts observed in the IR and ^{13}C NMR spectra for the 5- and 8-acetyl groups: 1695 cm $^{-1}$ (KBr) in IR and 215.5 and 215.7 p.p.m. ($CDCl_3$, Me_4Si) in ^{13}C NMR in comparison with 1710 cm $^{-1}$ and 208.7 p.p.m. for the 13-acetyl group.

Molecular packing in the crystal is shown in Fig. 2. The molecules are held together by van der Waals interactions with normal intermolecular contacts.

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Structures of an Antithrombotic Agent and its Dimethyl Derivative: 4-Methoxy- N,N' -bis(3-pyridylmethyl)isophthalamide Monohydrate (I) and 4-Methoxy- N,N' -dimethyl- N,N' -bis(3-pyridylmethyl)isophthalamide Hemihydrate (II)

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Abstract. (I): $C_{21}H_{20}N_4O_3 \cdot H_2O$, $M_r = 394.37$, monoclinic, $P2_1/c$, $a = 6.354$ (2), $b = 14.043$ (1), $c = 21.418$ (3) Å, $\beta = 92.33$ (2) $^\circ$, $V = 1909.9$ (6) Å 3 , $Z = 4$, $D_x = 1.37$ Mg m $^{-3}$, Mo $K\alpha$, $\lambda = 0.7107$ Å, $\mu = 0.058$ mm $^{-1}$, $F(000) = 832$, $T = 293$ K, $R = 0.066$ for 1830 observed reflections. (II): $C_{23}H_{24}N_4O_3 \cdot \frac{1}{2}H_2O$, $M_r = 413.41$, monoclinic, $P2_1/n$, $a = 31.649$ (4), $b = 11.753$ (2), $c = 11.501$ (2) Å, $\beta = 93.11$ (2) $^\circ$, $V =$

4272 (1) Å 3 , $Z = 8$, $D_x = 1.28$ Mg m $^{-3}$, Mo $K\alpha$, $\lambda = 0.7107$ Å, $\mu = 0.05$ mm $^{-1}$, $F(000) = 1752$, $T = 293$ K, $R = 0.065$ for 5369 observed reflections. A considerable change in the shape of the basic molecular skeleton is caused by the N -methylation in (II), an inactive compound, to be compared with the planar shape observed in (I), an antithrombotic agent (picotamide). The two independent molecules in (II) are substantially